

University of Groningen

Molecular modeling studies of lipase-catalyzed β -lactam polymerization

Baum, I.; Haller, L.A.; Schwab, L.W.; Loos, K.; Fels, G.

Published in:
 Chemistry Central Journal

DOI:
[10.1186/1752-153X-3-S1-P57](https://doi.org/10.1186/1752-153X-3-S1-P57)

IMPORTANT NOTE: You are advised to consult the publisher's version (publisher's PDF) if you wish to cite from it. Please check the document version below.

Document Version
 Publisher's PDF, also known as Version of record

Publication date:
 2009

[Link to publication in University of Groningen/UMCG research database](#)

Citation for published version (APA):

Baum, I., Haller, L. A., Schwab, L. W., Loos, K., & Fels, G. (2009). Molecular modeling studies of lipase-catalyzed β -lactam polymerization. *Chemistry Central Journal*, 3, [P57]. <https://doi.org/10.1186/1752-153X-3-S1-P57>

Copyright

Other than for strictly personal use, it is not permitted to download or to forward/distribute the text or part of it without the consent of the author(s) and/or copyright holder(s), unless the work is under an open content license (like Creative Commons).

The publication may also be distributed here under the terms of Article 25fa of the Dutch Copyright Act, indicated by the "Taverne" license. More information can be found on the University of Groningen website: <https://www.rug.nl/library/open-access/self-archiving-pure/taverne-amendment>.

Take-down policy

If you believe that this document breaches copyright please contact us providing details, and we will remove access to the work immediately and investigate your claim.

Downloaded from the University of Groningen/UMCG research database (Pure): <http://www.rug.nl/research/portal>. For technical reasons the number of authors shown on this cover page is limited to 10 maximum.

Poster presentation

Open Access

Molecular modeling studies of lipase-catalyzed β -lactam polymerization

I Baum^{*1}, LA Haller¹, LW Schwab², K Loos² and G Fels¹

Address: ¹Department of Chemistry, Faculty of Science, University of Paderborn, Paderborn, Germany and ²Department of Polymer Chemistry and Zernike Institute for Advanced Materials, University of Groningen, Groningen, The Netherlands

* Corresponding author

from 4th German Conference on Chemoinformatics
Goslar, Germany. 9–11 November 2008

Published: 5 June 2009

Chemistry Central Journal 2009, 3(Suppl 1):P57 doi:10.1186/1752-153X-3-S1-P57

This abstract is available from: <http://www.journal.chemistrycentral.com/content/3/S1/P57>

© 2009 Baum et al; licensee BioMed Central Ltd.

Enzymatic polymerization has emerged over the last 5 years as a field of considerable interest and commercial promise. The reaction proceeds with high regio-, enantio-, and chemoselectivity under relatively mild conditions. Enzymes have been used so far to synthesize polyesters, polysaccharides, polycarbonates, polyphenols, polyanilines, vinyl polymers, and poly-amino acids [1]. Particularly, lipase B of *Candida antarctica* immobilized on polyacrylic resin (Novozyme 435) has proven to be a very versatile catalyst and has successfully been used for the synthesis of polyesters from various substrates [2][3][4]. Little, however, has been reported on the enzyme catalyzed synthesis of polyamides [5].

While it has been shown that nylons can chemically be produced from the corresponding amino acids or by anionic ring-opening polymerization of 5–13 membered unsubstituted lactams, poly- β -alanine has not yet been obtained by either polymerization of β -alanine or β -lactam (2-azetidinone). Using lipase B of *Candida antarctica* we have recently been successful in the production of unbranched poly- β -alanine starting from unsubstituted β -lactam [6].

Here we report preliminary molecular modeling studies of the lipase catalyzed ringopening polymerization of β -lactam towards an understanding of the underlying enzymatic mechanism. We can show that amide formation initially follows the well-known enzymatic acylation of Ser105 by β -lactam using Asp187 and His224 of the catalytic centre and Thr40 and Gly106 as oxy-anion hole. The

elongation of the chain, however, utilizes different parts of the active site. The mechanism is only applicable for β -lactam and can not be utilized by β -alanine and suggests a reasoning for the experimental finding that β -alanine can not be polymerized enzymatically but rather inhibits the polymerization in a copolymerization experiment with β -lactam and β -alanine.

References

1. Kobayashi S, Ritter H, Kaplan D, eds: *Enzyme-Catalyzed Synthesis of Polymers (Advances in Polymer Science)* Springer, Berlin; 2007.
2. Thurecht KJ, Heise A, deGeus M, Villarroya S, Zhou JX, Wyatt MF, Howdle SM: *Macromolecules* 2006, **39**:7967.
3. Mee L van der, Helmich F, de Bruijn R, Vekemans JAJM, Palmans ARA, Meijer EW: *Macromolecules* 2006, **39**:5021.
4. Kumar A, Mei Y, Gross R: *Macromolecules* 2003, **36**:5530.
5. Gu Q-M, Maslanka VV, Cheng HN: *Polym Prepr* 2006, **4**:234.
6. Schwab LW, Kroon R, Schouten AJ, Loos K: *Macromol Rapid Commun* 2008, **29**:794-797.